

Studies of the EPR Parameters and the Local Tetragonal Distortion of V⁴⁺-doped SrTiO₃ Crystal

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The EPR parameters (g factors g_{\parallel} , g_{\perp} and hyperfine structure constants A_{\parallel} , A_{\perp}) of V⁴⁺ ion in the tetragonal phase of SrTiO₃ crystal were studied by high-order perturbation formulas based on a two-spin-orbit-parameter model, where the contributions from the spin-orbit-coupling-parameter of central 3d^{*n*} ion and that of ligands are considered. It is found that the oxygen octahedron surrounding the impurity ion V⁴⁺ changes from elongation along the tetragonal axis in the pure crystal to compression and the magnitude of tetragonal distortion of oxygen octahedron (characterized by $|\Delta R| = |R_{\parallel} - R_{\perp}|$) in V⁴⁺ center is much greater than the corresponding value in the pure crystal. The cause related to the strong Jahn-Teller effect is discussed.

Key words: Electron Paramagnetic Resonance; Crystal- and Ligand-field; Jahn-Teller Effect; V⁴⁺; SrTiO₃.

1. Introduction

SrTiO₃ has been of considerable theoretical and experimental interest because of its unusual properties [1]. The crystal has a perovskite structure which undergoes a structural phase transition from cubic to tetragonal at ≈ 105 K [1]. The tetragonal phase is the result of the rotation of TiO₆ octahedra around the tetragonal c -axes. The rotation of TiO₆ octahedra leads to slight elongation of the Ti-O bond along the c -axes due to acoustic-optic phonon coupling [2]. So, the TiO₆ octahedron in the tetragonal phase of SrTiO₃ is slightly elongated (i. e., the bonding length $R_{\parallel} > R_{\perp}$). This was confirmed by X-ray diffraction [3], high-angle double-crystal X-ray diffractometry (HADOX) [4] and neutron diffraction [5] for the pure SrTiO₃ crystal, and by EPR and optical spectra studies [6–8] for Cr³⁺-, Fe³⁺- and Mn⁴⁺-doped SrTiO₃ (where the impurity ions are used to be the probes, they occupy the Ti⁴⁺ sites and the charge compensators are not in the vicinity of impurity ions). It should be pointed out that the ground states of these impurity ions in octahedral sites are single orbit states, so there is no Jahn-Teller effect in these systems. Thus, only the small local tetragonal distortion may occur in these impurity centers (e.g. Cr³⁺ centers) [6] because

of the size and/or charge mismatch. However, if the impurity ion is a 3d¹ ion, e. g. V⁴⁺, since the ground state of 3d¹ ion in an elongated octahedron is an orbit doublet, the system is not stable. It can be expected that the Jahn-Teller effect may result in a large change of the (VO₆)⁸⁻ octahedron from elongation to compression, because in the case of a 3d¹ ion in a compressed octahedron the ground state is an orbit single state. The EPR spectra of V⁴⁺ in the tetragonal phase of SrTiO₃ ($T = 4.2$ K) were reported [9]. However, no theoretical calculations for these EPR parameters (g factors g_{\parallel} , g_{\perp} and hyperfine structure constants A_{\parallel} , A_{\perp}) have been made, and the sign and magnitude of the local tetragonal distortion of the oxygen octahedron (characterized by the value $\Delta R = R_{\parallel} - R_{\perp}$) have not been given. In order to confirm the above expectation, to explain the EPR parameters and to obtain the local structural data for SrTiO₃:V⁴⁺ system, in this paper, we study the EPR parameters by the high-order perturbation formulas based on the two-spin-orbit-parameter model. The results are discussed.

2. Calculation

Since the spin-orbit-coupling-parameter ($\zeta_p^0 \approx 150$ cm⁻¹ [10]) of the ligand O²⁻ is close to that

($\zeta_p^0 \approx 248 \text{ cm}^{-1}$ [11]) of the central $3d^1$ ion, we apply the high-order perturbation formulas of EPR parameters based on the two-spin-orbit-parameter model, where both the contribution to the EPR parameters from the spin-orbit-coupling-parameter of the central $3d^1$ ion and that of ligand ions are considered. Thus, the one-electron basis functions can be expressed as the linear combinations of the d orbitals $|d_r\rangle$ of the central $3d^n$ ion and the p orbitals $|p_r\rangle$ of the ligands, i. e.,

$$|\psi_\gamma\rangle = N_\gamma^{1/2}(|d_\gamma\rangle - \lambda_\gamma|p_\gamma\rangle), \quad (1)$$

$$\begin{aligned} g_{\parallel} &= g_s - 8k'\zeta'/E_1 - k\zeta^2/(E_2)^2 - 4k'\zeta\zeta'/(E_1E_2) - g_s\zeta^2/(E_2)^2, \\ g_{\perp} &= g_s - 2k\zeta/E_2 + k\zeta^2/(E_2)^2 - 2k\zeta'^2/(E_1E_2) + 2k'\zeta\zeta'/(E_1E_2) - 2g_s\zeta'^2/(E_1)^2 - g_s\zeta^2/2(E_2)^2, \\ A_{\parallel} &= P(-k - 4/7) + P' \left[-8k'\zeta'/E_1 - 6k\zeta/(7E_2) - 4k\zeta^2/(7E_2)^2 \right. \\ &\quad \left. - 6k\zeta'^2/(7E_1E_2) - 22k'\zeta\zeta'/(7E_1E_2) - 6g_s\zeta'^2/7(E_1)^2 - 17g_s\zeta^2/14(E_2)^2 \right], \quad (2) \\ A_{\perp} &= P(2/7 - k) + P' \left[-11k\zeta/(7E_2) + 11/14k\zeta^2/(E_2)^2 - 11k\zeta'^2/(7E_1E_2) \right. \\ &\quad \left. + 11k'\zeta\zeta'/(7E_1E_2) - 11/7g_s\zeta'^2/(E_1)^2 - 11/28g_s\zeta^2/(E_2)^2 \right], \end{aligned}$$

in which g_s (≈ 2.0023) is the spin-only value. κ is the core polarization constant. The energy separations E_i can be written as

$$\begin{aligned} E_1 &= E(^2B_1) - E(^2B_2) \approx 10Dq, \\ E_2 &= E(^2E) - E(^2B_2) = 5D_t - 3D_s, \end{aligned} \quad (3)$$

where Dq is the cubic field parameter, and D_s and D_t are the tetragonal field parameters.

The spin-orbit-parameters ζ , ζ' , the orbital reduction factors k , k' and the dipolar hyperfine structure parameters P , P' in (2) can be expressed as

$$\begin{aligned} \zeta &= N_t(\zeta_d^0 + \lambda_t^2\zeta_p^0/2), \\ \zeta' &= (N_tN_e)^{1/2}(\zeta_d^0 - \lambda_t\lambda_e\zeta_p^0/2), \\ k &= N_t[1 - 2\lambda_tS_{dp}(t_{2g}) + \lambda_t^2/2], \quad (4) \\ k' &= (N_tN_e)[1 - \lambda_tS_{dp}(t_{2g}) - \lambda_eS_{dp}(e_g) - \lambda_e\lambda_t/2], \\ P &= N_tP_0, \quad P' = (N_tN_e)^{1/2}P_0, \end{aligned}$$

where ζ_d^0 and ζ_p^0 are the spin-orbit-parameter of the d electrons of a free $3d^1$ ion and that of p electrons of a free ligand ion. P_0 is the dipolar hyperfine structure constant of the free $3d^1$ ion. For the studied $SrTiO_3:V^{4+}$ system, ζ_d^0 and ζ_p^0 are given above and $P_0 \approx 136 \cdot 10^{-4} \text{ cm}^{-1}$ [12, 13]. $S_{dp}(r)$ are the group overlap integrals. For $SrTiO_3:V^{4+}$, from the Slater-

where $\gamma = e_g$ or t_{2g} denotes the irreducible representation of the O_h group. N_γ and λ_γ are, respectively, the normalization factors and the orbital mixing coefficients.

From the g factor of $SrTiO_3:V^{4+}$, it can be expected that the ground state is 2B_2 ($|d_{xy}\rangle$). Thus, from the above one-electron basis function and the perturbation methods, the high-order perturbation formulas of $3d^1$ ion in the tetragonal symmetry can be derived as

type SCF functions [14, 15] and the metal-ligand distance $R_0 \approx 1.95 \text{ \AA}$ [3–5] we obtain $S_{dp}(e_g) \approx 0.1297$ and $S_{dp}(t_{2g}) \approx 0.0508$. In (2) we assume that the mixing coefficients $\lambda_\gamma = \lambda_e \approx \lambda_t$ and the coefficient λ_γ is taken as an adjustable parameter. Thus, the normalization factors N_r can be calculated from the normalization relation

$$N_\gamma[1 - 2\lambda_\gamma S_{dp}(\gamma) + \lambda_\gamma^2] = 1. \quad (5)$$

The tetragonal field parameters D_s and D_t can be calculated from the superposition model [16]. From the model, we have

$$\begin{aligned} D_s &= (4/7)\bar{A}_2(R_0)[(R_0/R_{\perp})^{t_2} - (R_0/R_{\parallel})^{t_2}], \\ D_t &= (16/21)\bar{A}_4(R_0)[(R_0/R_{\perp})^{t_4} - (R_0/R_{\parallel})^{t_4}], \end{aligned} \quad (6)$$

where the power-law exponents are $t_2 \approx 3$ and $t_4 \approx 5$ [17–19]. $\bar{A}_2(R_0)$ and $\bar{A}_4(R_0)$ are intrinsic parameters. For a $3d^n$ MX_6 octahedral cluster, $\bar{A}_4(R_0) = 3/4Dq$ [16, 18, 19] and $\bar{A}_2(R_0) \approx (9 \sim 12)\bar{A}_4(R_0)$ [18–21]. We take $\bar{A}_2(R_0) = 12\bar{A}_4(R_0)$ here. To our best knowledge, the value of Dq for the $(VO_6)^{8-}$ cluster is not reported, so we estimate it as follows: According to the spectrochemical series [22], for a $3d^n$ ion, $Dq(F^-) < Dq(O^{2-})$, for example, $Dq(F^-)$ in $KMgF_3:Cr^{3+}$ is about $1,560 \text{ cm}^{-1}$ [23] and $Dq(O^{2-})$

in Al₂O₃:Cr³⁺ is above 1,800 cm⁻¹ [24, 25]. Thus, from $Dq \approx 2,012$ cm⁻¹ for the (VF₆)²⁻ octahedral cluster [26], we estimate $Dq \approx 2,400$ cm⁻¹ for the studied (VO₄)⁸⁻ cluster. In the tetragonal phase of SrTiO₃:V⁴⁺ we have the bonding length $R_{\perp} \approx R_0 \approx 1.95$ Å [3–5], and the bonding length R_{\parallel} is taken as an adjustable parameter. The core polarization constant κ is related only to the hyperfine structure constants A_{\parallel} and A_{\perp} , it is also taken as the adjustable parameter. Thus, by fitting the calculated EPR parameters g_{\parallel} , g_{\perp} , A_{\parallel} and A_{\perp} to the observed values, we obtain for the tetragonal SrTiO₃:V⁴⁺ system

$$\lambda_{\gamma} \approx 0.4646, \quad R_{\parallel} \approx 1.859 \text{ Å}, \quad \kappa \approx 0.58. \quad (7)$$

Comparisons between the calculated EPR parameters and the experimental values are shown in Table 1.

Table 1. g Factors and hyperfine structure constants for V⁴⁺ in the tetragonal phase of SrTiO₃.

	g_{\parallel}	g_{\perp}	A_{\parallel} (10 ⁻⁴ cm ⁻¹)	A_{\perp} (10 ⁻⁴ cm ⁻¹)
Calculation	1.9421	1.8951	-146.76	-44.37
Experiment [9]	1.9420(50)	1.8945(50)	-146.78(5) ^a	-44.04(5) ^a

^a The sign is not given in [9].

3. Discussion

The above calculations suggest that the bonding length $R_{\parallel} < R_{\perp}$, that is to say, the oxygen octahedron in the V⁴⁺ center in the tetragonal SrTiO₃ crystal is compressed. This point is, as expected above,

opposite to that in the pure SrTiO₃ [3–5] and the Cr³⁺-, Fe³⁺- and Mn⁴⁺-doped SrTiO₃ crystals [6–8]. The magnitude of the tetragonal distortion $|\Delta R| = |R_{\parallel} - R_{\perp}|$ is 0.091 Å for the V⁴⁺ center in the tetragonal SrTiO₃. This value is much greater than the value ($\approx 5 \cdot 10^{-4}$ Å [3–5, 7, 8]) in pure SrTiO₃ and Fe³⁺- and Mn⁴⁺-doped SrTiO₃ at 4.2 K and that ($\approx 16 \cdot 10^{-4}$ Å [6]) in Cr³⁺-doped SrTiO₃. Considering that the impurity V⁴⁺ has the same charge as the replaced Ti⁴⁺ ion in SrTiO₃, tetragonal distortion due to charge compensation can be excluded. In addition, since the ionic radius (≈ 0.63 Å [27]) of V⁴⁺ is close to that (≈ 0.68 Å [27]) of the replaced Ti⁴⁺, one has to visualize that off-center displacement of the impurity is the cause of the large tetragonal distortion. So, the large tetragonal distortion of the V⁴⁺ center in the tetragonal phase of SrTiO₃ is attributed to the Jahn-Teller effect, because V⁴⁺ in SrTiO₃ is a Jahn-Teller system. It can be expected that even for V⁴⁺ in cubic SrTiO₃, the Jahn-Teller effect may result in a tetragonal elongation of the oxygen octahedron. Noteworthy, for the 3d¹ ions Ti³⁺ and Cr⁵⁺ in the tetragonal phase of SrTiO₃ [28, 29], since $\bar{R}_{\parallel} > R_{\perp}$, the ground state is an orbital doublet, an additional distortion due to the Jahn-Teller effect and/or the off-center displacement of an impurity ion (only for Cr⁵⁺) can occur, which leads the symmetry of the octahedron being orthorhombic and the orbital degeneracy of the doublet is removed. So, if a 3d¹ ion is in an octahedral site with tetragonal distortion, the tetragonal distortion should be compressed and even the corresponding octahedron in the host crystal is elongated.

- [1] M. E. Lines and A. M. Glass, Principles and Applications of Ferroelectrics and Related Materials, Oxford University Press, Oxford / London 1977.
- [2] H. Thomas and K. A. Müller, Phys. Rev. Lett. **21**, 1256 (1968).
- [3] M. Hidaka, S. Maeda, and J. S. Storey, Phase Transitions **5**, 219 (1985).
- [4] M. Sato, Y. Soejima, N. Ohama, A. Okazaki, H. J. Scheel, and K. A. Müller, Phase Transitions **5**, 207 (1985).
- [5] B. Alefeld, Z. Phys. **222**, 155 (1969).
- [6] W. C. Zheng, J. Phys.: Condens. Matter **6**, 2241 (1994).
- [7] W. C. Zheng, Solid State Commun. **89**, 467 (1994).
- [8] W. C. Zheng, Phys. Status Solidi B **184**, K11 (1994).
- [9] Th. W. Kool and M. Glasbeek, Solid State Commun. **32**, 1099 (1979).
- [10] M. L. Du and C. Rudowicz, Phys. Rev. B **46**, 8974 (1992).
- [11] J. S. Griffith, The Theory of Transition-metal Ions, Cambridge University Press, London 1964.
- [12] R. Biyik, R. Tapramaz, and B. Karabulut, Z. Naturforsch. **58a**, 499 (2003).
- [13] L. J. Boucher, E. C. Jynan, and T. F. Yen, Electron Spin Resonance of Metal Complexes, Plenum Press, New York 1969.
- [14] E. Clementi and D. L. Raimondi, J. Chem. Phys. **38**, 2686 (1963).
- [15] E. Clementi, D. L. Raimondi, and W. P. Reinhardt, J. Chem. Phys. **47**, 1300 (1967).
- [16] D. J. Newman and B. Ng, Rep. Prog. Phys. **52**, 699 (1989).
- [17] W. L. Yu, J. Phys.: Condens. Matter **6**, 5105 (1994).
- [18] W. C. Zheng, S. Y. Wu, and J. Zi, Z. Naturforsch. **56a**, 473 (2001).
- [19] W. C. Zheng and S. Y. Wu, Spectrochim. Acta A **57**, 1177 (2001).

- [20] C. Rudowicz and Y. Y. Zhou, *J. Magn. Magn. Mater.* **111**, 153 (1992).
- [21] A. Edgar, *J. Phys.* **C9**, 4304 (1976).
- [22] J. A. Lever, *Inorganic Electronic Spectroscopy*, Elsevier Press, Amsterdam 1984.
- [23] N. S. Altshuler and A. L. Lanionov, *Opt. Spectrosc.* **66**, 61 (1989).
- [24] D. S. McClure, *J. Chem. Phys.* **36**, 2757 (1962).
- [25] D. S. McClure, *J. Chem. Phys.* **38**, 2289 (1963).
- [26] H. D. Bedon, W. E. Hatfield, S. M. Horner, and S. Y. Tyree Jr., *Inorg. Chem.* **4**, 743 (1965).
- [27] R. C. Weast, *CRC Handbook of Chemistry and Physics*, CRC Press, Boca Raton 1989, p. F-164.
- [28] V. V. Laguta, M. D. Glinchuk, R. O. Kuzian, S. N. Nokhrin, I. P. Bykov, J. Rosa, L. Jastrabik, and M. G. Karkut, *J. Phys.: Condens. Matter* **14**, 13813 (2002).
- [29] Th. W. Kool, H. J. de Jong, and M. Glasbeek, *J. Phys.: Condens. Matter* **6**, 1571 (1994).